



Development and validation of a detailed NH₃ combustion mechanism CEU-NH₃

Zhihua Wang^{1,*}, Xinlu Han², Shixing Wang³

*lead presenter: wangzh@zju.edu.cn

1 State Key Laboratory of Clean Energy Utilization, Zhejiang University, Hangzhou, 310027, PR. China.

2 College of New Energy, China University of Petroleum (East China), Qingdao, 266580, PR. China.

3 Clean Combustion Research Center, King Abdullah University of Science and Technology, Thuwal, Saudi Arabia

The usage of fossil fuels like coal, oil, and natural gas has caused serious challenges to human societies with pollution emissions as well as greenhouse gases like CO₂. According to the data of IEA, the global CO₂ emission in 2023 has reached 37.4 billion tonnes (Gt), with a 1.1% increase ratio^[1]. More than 198 countries have signed the 2015 Paris Agreement. To control the global average temperature rise limited to 1.5 °C, it's urgently needed by all countries switching fossil fuel-based energy systems to a renewable and sustainable one. As one of the most important zero-carbon fuels, ammonia is playing a more and more important role in the combustion area and is considered to replace typical fossil fuels^[2]. The green ammonia can be synthesized through the Haber-Bosch process with green H₂ and N₂. Without carbon in the molecule, and easy for liquification it's one of the best H₂ carriers. Ammonia can be directly or co-fired with natural gas, oil, and coal in the existing gas turbines, engines as well as industrial boilers and furnaces.

The combustion of ammonia is a challenge due to low reactivity and risk of extremely high NO_x emission. Developing and validating of NH₃ combustion mechanism is essential for understanding the reaction process and developing a new type of combustion technology for NH₃ fuels. In this paper, the fundamental experiments based on heat flux methods and a detailed NH₃ combustion mechanism of CEU-NH₃ will be introduced^[3,4]. Based on the kinetic simulation platform of CHEMKIN, the mechanism will be validated by a variety of experiment data like laminar burning velocity, ignition delay time, and species concentration obtained from experimental data for pure NH₃, NH₃/H₂, NH₃/CH₄, NH₃/Syngas, and NH₃/Methanol/Ethanol. A temperature dependence parameter α and pressure dependence parameter β model will be introduced and discussed. The over-rich phenomenon of different fuels will be discussed as well. The NO_x formation route and control strategy will be discussed based on kinetic simulation results. The sensitivity analysis and rate of production (ROP) for key species will be analyzed.

Acknowledgments: This work was funded by the National Key R&D Program of China (2022YFB4003902) and the National Natural Science Foundation of China (52125605).

References

- [1] <https://www.iea.org/reports/co2-emissions-in-2023/executive-summary>.
- [2] Valera-Medina A, Xiao H, Owen-Jones M et al. Ammonia for Power. Progress in Energy and Combustion Science, 2018; 69: 63-102.
- [3] Han X, Wang Z, Costa M, et al. Experimental and kinetic modeling study of laminar burning velocities of NH₃/air, NH₃/H₂/air, NH₃/CO/air and NH₃/CH₄/air premixed flames. Combustion and Flame. 2019; 206: 214-226.
- [4] Wang S, Wang Z, Chen C, et al. Applying heat flux method to laminar burning velocity measurements of NH₃/CH₄/air at elevated pressures and kinetic modeling study. Combustion and Flame. 2022; 236: 111788.