

Growth of NaBaCr₂(PO₄)₃ crystals by high temperature solution method

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Frameworks built with transition metals MO₆ octahedra and (XO₄)ⁿ⁻ polyanions (X = phosphorus, chalcogen, silicon, aluminium or some 3d and 5d transition metal) offer the same kind of chemical flexibility that the well-known perovskite and/or spinel-type structures. Fundamental interest in polyanionic frameworks regarding magnetic properties appears quite new. Using (XO₄)ⁿ⁻ polyanions as a building element, rather than the more traditional oxo anion, will help to obtain ferroic solids. NaBa_{1-x}Sr_xCr₂(PO₄)₃ solids, ceramics, with x=0 and 1 were recently reported to show intriguing physical properties (hysteric magneto capacitance signal when x=0 and a parallel magnetic long-range ordering when x=1) [1]. In order to investigate the physical properties / nuclear relationship with respect to an external stimulus single crystals of NaBa_{1-x}Sr_xCr₂(PO₄)₃ with Langbeinite type structure are needed.

Several Langbeinite type crystals (*P*2₁3 space group) have already been obtained by high temperature solution growth in molten phosphate salts but single crystals of NaBa_{1-x}Sr_xCr₂(PO₄)₃ have never been obtained by this method [2,3]. In this work we study three systems (Na₂O – NaBaCr₂(PO₄)₃ – NaPO₃ & RbF – NaBaCr₂(PO₄)₃ – RbPO₃ & Na₂O – NaBaCr₂(PO₄)₃ – WO₃) in order to determine if the crystals of interest can be grown. In all the three systems NaBaCr₂(PO₄)₃ crystals have been obtained by spontaneous nucleation but in phosphate fluxes a compositional change occurs during the crystal growth process due to a high evaporation of the solution. In molten tungstate salts no evaporation of the solution is observed but tiny crystals belonging to a secondary phase (Cr₂WO₆) have been obtained (Fig. 1).

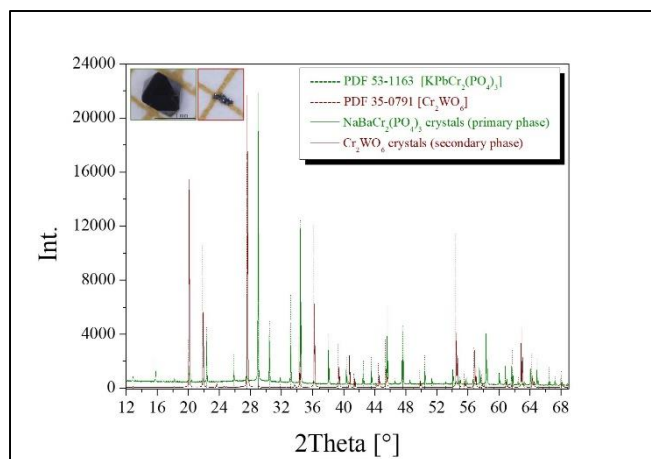


Figure 1. Powder x-ray diffraction of the crystals obtained by spontaneous nucleation from a high temperature solution of the chemical the system Na₂O – NaBaCr₂(PO₄)₃ – WO₃. A picture of one NaBaCr₂(PO₄)₃ crystal (left hand side) and several Cr₂WO₆ crystals (right hand side) are shown in the inset.

Solutions with different compositions of the system Na₂O – NaBaCr₂(PO₄)₃ – WO₃ are under study in order to determine the compositional zone where only NaBaCr₂(PO₄)₃ crystals are stable.

[1] Souiwa K et al. Synthesis and characterization of the phosphates Na_{1+x}Mg_{1-x}Cr_{2-x}(PO₄)₃ (x = 0; 0.2) and NaZnCr₂(PO₄)₃ with the α -CrPO₄ structure. J Alloys Compd. 2015;627:153-160.

[2] Carvajal JJ et al. Growth and structural characterization of Rb₂Ti_{1.01}Er_{0.99}(PO₄)₃. Chem. Mater. 2003;15:204-211.

[3] Peña A et al. Yb:Ta:RbTiOPO₄, a new strategy to further increase the lanthanide concentration in crystals of the KTiOPO₄. Chem. Mater. 2007;19:4069-4076.