

Salts of amino acids with a dimeric cation of the $[A(1)^+\cdots A(2)]$ type with polar symmetry

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Recently, we discovered a new class of salts containing different amino acids. The vast majority (47) of them are of a type $[A(1)H\cdots A(2)]X$, where A(1) is protonated and A(2) zwitterionic amino acid: glycine (Gly), sarcosine (Sar), dimethylglycine (DMG), betaine (Bet), β -alanine (β -Ala), L-proline (L-Pro). X stands for Cl, Br, I, NO₃, BF₄, ClO₄ and NH₂SO₃ anions. The following dimeric cations were established: (GlyH \cdots Sar), (GlyH \cdots DMG), (SarH \cdots DMG), (DMGH \cdots Sar), (SarH \cdots Bet), (BetH \cdots Sar), (DMGH \cdots Bet), (BetH \cdots DMG), (L-ProH \cdots Sar), (β -AlaH \cdots Sar), (β -AlaH \cdots DMG), (β -AlaH \cdots Bet) and (β -AlaH \cdots L-Pro). Sixteen of these salts crystallize with polar symmetry: [(SarH \cdots DMG)BF₄, (SarH \cdots DMG)ClO₄ - space group *Pc*], [(DMGH \cdots Sar)ClO₄ - s. g. *Pna2₁*], [(L-ProH \cdots Sar)BF₄, (L-ProH \cdots Sar)ClO₄, (L-ProH \cdots Sar)NO₃, (L-ProH \cdots Sar)Br, (L-ProH \cdots Sar)I - s. g. *C2*], [(β -AlaH \cdots DMG)I, (β -AlaH \cdots DMG)BF₄, (β -AlaH \cdots DMG)NH₂SO₃ - s. g. *Pna2₁*], [(β -AlaH \cdots Bet)Br, (β -AlaH \cdots Bet)I - s. g. *Cc*], [(β -AlaH \cdots L-Pro)Cl, (β -AlaH \cdots L-Pro)Br, (β -AlaH \cdots L-Pro)I - s. g. *P2₁*]. The last three salts were described in our recently published article [1]. All these salts are potential pyroelectric or ferroelectric materials, the same as the recently reported ferroelectric (DMGH \cdots DMG)Cl [2], as well as candidates for piezoelectric and nonlinear optics applications. Protonation in a pair of amino acids may depend on the anion, as in the cases of (SarH \cdots Bet)ClO₄ and (BetH \cdots Sar)NH₂SO₃·0.5H₂O, (DMGH \cdots Bet)ClO₄·H₂O and (BetH \cdots DMG)I·H₂O or, for the same anion, on the crystallization conditions, as in the case of (SarH \cdots DMG)ClO₄ (crystallized from water solution) and (DMGH \cdots Sar)ClO₄ (crystallized from acetic acid).

Work on the study of the properties of these salts is in progress.

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References

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- [2] Szafranski M et al. Above-room-temperature ferroelectricity and piezoelectric activity of dimethylglycinium-dimethylglycine chloride. *Mater Des* 2022; 220:110893.