

Impact of chemistry on the interface with substrate of MoS₂ nanosheets grown by ambient pressure chemical vapor deposition

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Few layered MoS₂ attracts much of interest in different fields thanks to their peculiar properties. When considering micro and nano electronics, MoS₂ nanosheets are foreseen for integration as a channel material for overcoming the short-channel effects in ultra-scaled field-effect transistors (FETs).[1][2] However, the on-current in such short-channel FETs is limited by the contact resistance (R_c) related to energy barrier at the interface between the semiconductor channel and contact metal. Furthermore, ultrathin MoS₂ is also used in tunneling stacks to fabricate transistors with very steep sub-threshold slope, mandatory for low voltage electronics.[3] In these devices, the electron band alignment at MoS₂ interfaces directly affects the electrostatics of the stacks such as built-in voltages, transistor thresholds, as well as tunneling barrier heights. In this respect, a detailed study of the interface between the MoS₂ nanosheets and the substrate is essential to unveil in details its properties and possibly correlated them with the specific conditions during the growth process. When chemical-based growth methods are employed to obtain the MoS₂ nanosheets, the impact of the chemistry used in the precondition of the substrate and during the growth process needs a dedicated investigation to discriminate any role in defining the properties at the MoS₂/SiO₂ interface.

On this scope we grew by atmospheric pressure chemical vapor deposition (AP-CVD) 2 to 3 layers MoS₂ nanosheets on SiO₂(50nm)/Si using different methods based on vapor phase reaction (VPR) of MoO₃ and sulfur solid powders or sulfurization of solid precursor ultrathin film (SPF), spin-coated, containing Mo compounds (eg. AHM) in solution with Na or K based chemicals, such as NaOH or KCl. We characterize the so-grown MoS₂ with Raman spectroscopy, SEM, AFM, XPS and internal photoemission of electrons (IPE). The latter, that determines the energy of the semiconductor valence band (VB) relative to the reference level of the insulator conduction band (CB), unveiled a significant change (≈ 0.6 eV) in the electron barrier depending on the growth method, possibly ascribed to the interaction of hydroxile groups from NaOH and AHM promoters with the oxide surface leading to interface dipole modification, in SPF approach.

References (if needed)

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