

X-ray quantitative investigation of *in situ* grown 2D Transition metal dichalcogenide TiS₂ prepared by hybrid ALD/MLD

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Transition metal dichalcogenides (TMD) are receiving great interest in the past few years due to their future applications in super-capacitors, batteries, electronics and optoelectronics etc [1]. Lamellar titanium disulfide TiS₂, which consists of S-Ti-S layers separated by van der Waals gaps, is also considered to be integrated into emerging energy devices such as rechargeable batteries [2]. Ultra-thin films of TiS₂ are equally technologically challenging to prepare in large-scale production for device application. Among the various growth methods, atomic layer deposition (ALD) is the technique that could produce a high-quality ultra-thin film of TiS₂ in a controlled manner [3]. Recently, our group has demonstrated the controlled growth of TiS₂ ultra-thin films by a two-step process composed of (i) ALD/MLD step using inorganic and organic precursors (tetrakis-dimethylamido titanium and 1,2-ethanedithiol); (ii) annealing under Ar/H₂ atmosphere [4]. This work aims to understand both steps, i.e. the reaction of precursors with the substrate and subsequent growth cycles using *in situ* monitoring by X-ray. The *in situ* growth of ultra-thin films has been carried out on thermal SiO₂ on Si substrate in a custom-built portable reactor designed to be installed on the 6-axis diffractometer of beamline SIRIUS at SOLEIL Synchrotron [4]. X-ray fluorescence (XRF), X-ray reflectivity (XRR) and X-ray absorption spectroscopy (XAS) have been performed *in situ* during the sample growth and the subsequent annealing, to understand and control the growth mechanism. The XAS measurements have been performed at both the Ti and S K-edges.

To understand the initial growth process, Density Functional Theory (DFT) calculations have been performed to obtain the best suitable model structure. The DFT-optimized structures are used to simulate the XANES spectra, and the initial model is revised until the best matching. The analysis gives an atomistic view of the mechanism underlying the initial growth cycles with a quantitative approach (*in situ* XAS and DFT). The quantitative analysis of S K-edge data gives stronger evidence of the formation of TiS₂ at the final stage.

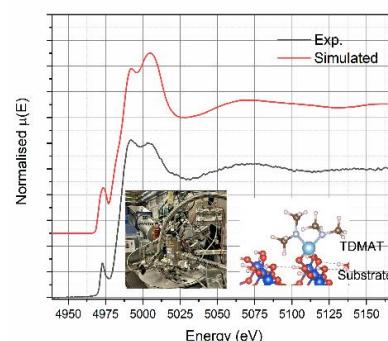


Fig: Experimental and simulated XANES spectra at Ti K-edge for the first half cycle. Insets show experimental setup and possible bonding with surface.

[1] Flamary-Mespoulie et al., Lithium-rich layered titanium sulfides: Cobalt- and Nickel-free high capacity cathode materials for lithium-ion batteries, *Energy Storage Mater.* 2020; 26:213–222.

[2] V. Pore et al., Atomic layer deposition of titanium disulfide thin films, *Chem. Vap. Depos.* 2007;13:163–168.

[3] M. Mattinen et al., Atomic Layer Deposition of 2D Metal Dichalcogenides for Electronics, Catalysis, Energy Storage, and Beyond, *Advanced Materials Interfaces* 2021; 8 (6): 2001677.

[4] P. A. Younes et al., Transition Metal Dichalcogenide TiS₂ Prepared by Hybrid ALD/MLD: Atomic-Level Insights with In Situ Synchrotron Xray Studies and Molecular Surface Chemistry, *Chem.Mater.* 2022;34: 10885.