

Role of N Diffusion Kinetics in Creating an Atomically Chemically Graded Ti/TiN Interface

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ABSTRACT

Recent calculations based on DFT suggest that it is thermodynamically favorable for N to diffuse across a sharp Ti/TiN interface from TiN to Ti [1]. Only migration of N atom across a sharp Ti/TiN interface is required to create the atomically chemical-graded interface, as a range of Ti-N stoichiometry could be achieved by migration of N without changing Ti position. On TiN side of the interface, Ti₆N₅, Ti₄N₃, Ti₃N₂ can form from rock salt TiN by forming ordered N vacancy. Similarly, on Ti side, N interstitials solid solution in Ti can form by intercalation of N. We explore the diffusion path of N migration from TiN to Ti and the rate of N diffusion. We calculate the diffusion coefficient of N in: 1) TiN, 2) Ti₆N₅, 3) Ti₄N₃, 4) Ti, 5) Ti-0.14N 6) Ti-0.2N, and 7) across the sharp Ti/TiN interface. We use DFT to calculate the barrier to diffusion of N along various possible paths and frequencies at the stable and transition states. Then, we calculate the diffusion coefficient within Arrhenius theory [2]. We show that the diffusion coefficient for N diffusion is slowest in TiN and fastest in Ti, while that across the interface is between TiN and Ti. Slowest diffusion of N in TiN leads to the formation of the chemically graded interface on annealing alone.

Keywords: Metal/ceramic interface; vacancy formation energy; interstitial formation energy; barrier energy; diffusion coefficient; Arrhenius theory density functional theory; nanoheterostructure

References

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